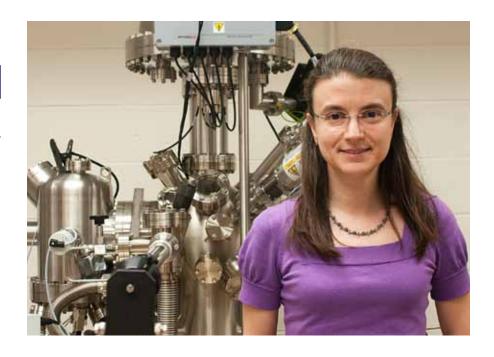
Department of Nuclear Science and Engineering Massachusetts Institute of Technology

## RESEARCH

## Better Fuel Cells and Reactors, Built From the Atomic Level



aterials are one of the key enablers in any engineering project. Bridge builders, who once worked in stone, and then in steel, can now draw on carbon fiber, polymers, and composites with outstanding strength, weight and other characteristics that extend the engineer's capabilities.

Engineers developing advanced energy technologies like fuel cells and new-generation reactors will benefit from novel materials knowledge being developed at NSE's Laboratory for Electrochemical Interfaces, headed by Assistant Professor Bilge Yildiz. Researchers there are working towards better understanding of the interfacial properties of oxides and learning to tailor their levels of ion transport, which could help create components with increased resistance to corrosion and other specialized properties.

To learn more about NSE please contact Professor Richard K. Lester, Head Department of Nuclear Science & Engineering rklester@mit.edu An especially exciting area of inquiry at the Laboratory for Electrochemical Interfaces is the use of lattice strain (a slight displacement of the atoms in a material's lattice structure) to alter the properties and boost the performance of oxides subjected to high temperature and reactive environments both in fuel cells and in nuclear applications.

"The purpose is to assess how strain can alter the efficiency of devices like fuel cells, or the lifetime of structural materials in reactors. We investigate how critical properties, like diffusivity of oxide ions or surface reactivity, can be controlled by strain," explains Yildiz, principal investigator at the Laboratory, which has received funding from an MIT Energy Initiative Grant, the Department of Energy's Office of Basic Energy Sciences, and a Nuclear Regulatory Commission Faculty Development seed grant.

Yildiz cites the case of strained yttria-stabilized zirconia as a recent example. The oxide-based ceramic material is used in solid-oxide fuel cells as an electrolyte, and also in reactors as a protective native film for cladding on nuclear fuel. Yildiz and postdoctoral researcher Akihiro Kushima developed the first direct proof that a precisely controlled and directed amount of lattice strain could increase the material's oxide ion conductivity by almost four orders of magnitude at 400 degrees K.

This property, the rate at which oxygen ions will diffuse through the material, is a critical engineering factor, notes Yildiz. "In fuel cells, you want the ions to diffuse quickly, so you're looking for the fastest strain. When you want to prevent corrosion in a reactor, you want them to diffuse slowly. We're studying different compositions for different functions."

"We're developing a smaller-scale understanding of how to control material properties, which hopefully will contribute to a larger-scale enhancement of energy technology performance."

In addition to exploring the mechanistic effects of strain on material interfaces, the Laboratory is also probing surfaces of fuel cell electrodes and of passive films in corrosion in their reactive environments, using an array of newly developed analytical and metrology equipment. This includes a one-of-a-kind scanning tunneling electron microscope capable of structural and electronic interrogation at temperatures of 700 degrees C, coupled with an X-ray photoelectron spectrometer for chemical analysis.

By developing this molecular-level understanding of critical materials, Yildiz and her associates hope to enable new levels of control and planning for future generations of fuel cells and reactors. The findings could enable solid-oxide fuel cells to operate at lower temperatures for better economic viability, or help designers improve the corrosion resistance of structures in nuclear plants and make more accurate estimates of their lifetimes.

As Yildiz puts it, "We're developing a smaller-scale understanding of how to control material properties, which hopefully will contribute to a larger-scale enhancement of energy technology performance."

*In the background of the image:* 

A surface science system including a variable temperature scanning tunneling microscope (STM), non-contact atomic force microscope (nc-AFM), and X-ray photoelectron spectrometer. The system enables to perform STM/nc-AFM measurements at elevated temperatures and reactive gas environments.

Written by Peter Dunn . Photo by Andrea Robles